

Norwegian University of Life Sciences





for Water Research



CENTRE FOR ENVIRONMENTAL RADIOACTIVITY



Norwegian Meteorological Institute







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Radioactive particles in the environment – sources and potential impact Brit Salbu CERAD CoE Environmental Radioactivity Norwegian University of Life Sciences (UMB)







Take home messages

- When refractory radionuclides (U, Pu) are released during nuclear event, particles ranging from submicrons to fragments should be expected
- The source determines the composition, the release scenarios dictate particle properties
- Radioactive particles can act as point sources
- Ignoring particles, the overall uncertainties in impact and risk assessments are significant
- Advanced techniques needed to characterise particles



A series of sources have contributed to releases of radioactive particles – Red: NMBU expeditions



- Nuclear weapon tests (Kazakhstan)
- Conventional detonation of weapons (Greenland, Spain)
- Nuclear reactor explosions and fires (Ukraine, UK, Canada)
- Accidents with reactor driven vehicles: satellites, submarine accidents (Russia, Norway)
- Effluents from nuclear installations (UK, France, USA, Russia, Sweden)
- Leaching from dumped nuclear material (Kara Sea, Barents Sea)
- Uranium mining and tailing (Central Asia, Norway)
- Use of DU ammunition (Kosovo, Kuwait)

Releases: Radioactive particles containing a series of radionuclides and metals



Radioactive particles released during "all" types of severe nuclear events. The source determines the composition, the release scenarios dictate particle properties

Nuclear test Semipalatinsk





Particle deposition

Hot spots – problems with representative sampling
Partial leaching – analytical errors - transuranics
May underestimate the inventories

te from the oyl explosion

Adds significantly to the overall uncertainties

Thule

Corrosion product Waste in Kara Sea

Krasnoyarsk U particle

Radioactive particles from soils - Krasnoyarsk Sample splitting combined with γ-spectrometry





Bulk (minus particle): ~100 g ~40 counts per second (Nal detector)

Sample splitting

Isolated grains of soil incl. particle: mg ~60 000 cps (Nal detector) -99,95% 436 000 Bq ¹³⁷Cs



Definition: Particles (IAEA, 2011)



Radioactive particles in the environment: are defined as localised aggregates of radioactive atoms that give rise to inhomogeneous distribution of radionuclides significantly different from that of the matrix background

In water/sediment/soil/biota

- Fragments: 2mm
- Particles : size range 0.45 µm 2mm
- Colloids /nanoparticles: size range: 1 nm 0.45 µm
- Low molecular mass species: less than 1 nm

In air: classified according to their aerodynamic diameters, less than 10 μ m are considered respiratory.



Advanced techniques available for particle characterization – state-of-the-art

- Hot spots/heterogeneities: digital autoradiography and sample splitting gamma measurements
- Particle size, surface structure and elemental composition: ESEM with XRMA, TEM with XRMA
- Subsurface/volume elemental composition: SR-based 2D µ-XRF (fluorescens)
- Oxidation state determination: SR-based 2D µ-XANES (micro X-ray absorption near edge structure spectrometry
- Crystallographic structure: SR-based µ-XRD (micro X-ray diffraction)
- 3D elemental distribution: Confocal µ-XRF, TOF-SIMS
- 3D structure distribution: Tomographic µ-XRD
- Source identification: Isotope or atom ratios by MS techniques (ICP-MS, AMS)
- Weathering and mobilisation potential: Leaching experiments



Source: uranium in soils (DU Kuwait) Info: Particle size distributions, surface elemental distributions, identifying single U particles: Autoradiography – SEM/BEI mode



SEM - XRMA Localization, Isolation and Characterization of U containing particles



SE, 255



UMa1,SE, 255



SEM and XRMA

- SEI-mode
 - characterization of particle surface structure.
- **BEI-mode**

 localization of particles containing heavy elements

- X-ray mapping localization of
 - particles containing radionuclides.

XRMA

element analysis

В X-ray Absorption Spectroscopy - ESRF Storage Ring Pre-Edge Region Accelerators Experimental Hall XANES EXAFS Photons -> exitation -> •µ-XRF- element deexitation **Edge Region** µ-XANES-ox state K, L, M electrons •µ-XRD-structure Emittance •EXAFS – fine structure X-ray fluorescence (XRF) Auger electrons Transmission of the ·50 50 150 200 250 300 100 1000 0 beam – absorption Relative Energy (eV)

Experimental set-up for: μ-XRF, μ-XAS, μ-XRD, μ-XANES, ESRF





XAS microtomography of a U fuel particles SEM and SR based reconstructed μ -tomography (ESRF), computerized slicing





SEI, 2000X

XAS TOMOGRAPHY



Setup for: μ -XRF, μ -XAS, μ -XRD, μ -XANES, HASYLAB.







Source: Chernobyl particles containing a series of radionuclides 2D nad 3D elemental distribution: Synchrotron radiationµ-XRF mapping





Chernobyl particles: inclusion of Ru+Mo Corresponding distributions: U, Zr, Sr



Synchrotron radiation μ -XANES for determination of oxidation states of U in a Chernobyl U particle (ESRF) – should be combined with μ -XRD.



U

I M +

INFO: 2 and 3D element and structure distribution by nanotomography (lab based absorption tomography) 3D video











RADIOACTIVE PARTICLES

NUCLEAR EXPLOSIONS AND SAFETY TESTS

543 atmospheric nuclear weapon tests



All radionuclides (except 3H, 14C and the long lived rare gases) involved in a nuclear detonation are accounted for completely as radioactive particles (Heft, 1970)



Particle from Nevada test site (Crocker et al., 1966)



Peaceful underground nuclear explosions in Russia (AMAP, 1998)

Tests of nuclear weapons in the atmosphere and underground (UNSCEAR, 2000)





Case: 50 Mt atmosperic nuclear device, Novaya Zemlya test site, October 30, 1961









Estimated ground deposition of Cs-137 from nuclearweapon fallout, decay converted for 1995.



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Isotope ratio fingerprinting – nuclear forensic

- Pu (and U) atom ratios vary with:
 - Type of reactor/fuel
 - fuel burn-up time
 - flux and energy
 - nuclear detonations: weapon type and yield
- \rightarrow tool for identification of source
- Global fallout (²⁴⁰Pu/²³⁹Pu: 0.17 – 0.19)
- Weapon grade Pu sources (²⁴⁰Pu/²³⁹Pu < 0.07)
- Tropospheric sources (low yield) (²⁴⁰Pu/²³⁹Pu ~0.04)



0.100

²³⁵U/²³⁸U atom ratio

1,000

10,000

100,000

1,0E+00

0,001

0,010



Source identification: Pu atom ratios in particles (AMS)



Source: air filters collected in Norway in October 1962 (Wendel et al, 2013). INFO: Autoradiography images - Pu and U isotope ratios (AMS) Hot spots still visible by autoradiography







Looking for the releasing source: Pu isotope ratio, air transport

NOAA HYSPLIT MODEL

Cencentration (/m3) averaged between 0 m and 6000 m Integrated from 0900 31 Oct to 2100 31 Oct 62 (UTC) TEST Release started at 0900 31 Oct 62 (UTC)



NOAA HYSPLIT MODEL

Concentration (/m3) averaged between 0 m and 75 m Integrated from 2200 09 Nov to 2300 09 Nov 62 (UTC) SP Release started at 0900 31 Oct 62 (UTC)



U

Source: Particles from underground nuclear weapon tests at Semipalatinsk Test Site (STS)

During 1949 -1989: 456 nuclear tests: 116 atmospheric tests and •340 underground tests were conducted

Norweg of Life S



•At Degelen Mountain nuclear test underground facility 224 tests were conducted (1961-1989)







Mayak PA field work 1994, 1996, Evaluation of potential accidents 2000



Contaminated Mayak water reservoirs in the river Techa





Figure 2.5. The industrial reservoirs in the upper part of the Techa

Sample collection, Reservoir 10

NTRE FOR ENVIRONMENTAL RADIOACTIVIT

Nuclear forensic: identifying sources from fingerprints: Isotope ratios for Pu-isotopes reflect military and civil sources





CHERNOBYL FUEL PARTICLES

West



0 10 20 30 40 50 60 70 μm





U

XANES AND XRD results:

- West (explosion): non-oxidised or even reduced forms of U
- North (fire): UO₂ core surrounded by oxidised U (U₂O₅/U₃O₈ layer)

→ Same source – different release conditions

CASE: The Chernobyl accident: 3-4 tons of U fuel released





Particles from Chernobyl

- At least 5 different types of particles:
 - Inert U-O-Zr, U- O-cabide
 - $-UO_2$
 - Oxidized U: e.g., U_3O_8
 - Single element particles (e.g.Ru)
 - Condennsed particles: volitiles deposited on surfaces of non-nuclear meterisl (fly ash)
- Particle characteristics vary according to composition, size distribution, crystallographic structure, porosity, oxidation state of the carrying paticle matrix (U)
- Particle weathering and remobilization of associated radionuclides depend on soil pH and abiotic and biotic oxidation (microbial activity)
- Particles represent a delay in the ecosystem transfer
- Particles may act as a diffuse source in the future





Bioerosion by hyphens: increased particle weathering and mobilisation of associated radionuclides - kinetics



Hypha of Piloderma croceum attached to a crystalline nuclear fuel particle collected within the 30 km exclusion zone around ChNPP, bar 5 μm



Norm particles



Soil from Kurday, Kazakhstan



Soil from Kadji-say, Kyrgyzstan



U contamination, Kadji-say, Kyrgyzstan



Heterogeneous TENORM particles from U mining in Kadji-say – XRD_



→uraninite $(UO_2 \text{ with} U(IV))$ coexisting with Na-zippeite $(Na_4(UO_2)_6[(OH)_{10}(SO_4)_3] \cdot 4H_2O)$

•Solubility of Na-zippeite (log Ksp= -116.5) < uraninite (UO₂; log K_{sp}= - 60.6)

•Na-zippeite: secondary alteration product of uraninite in oxidized zone of U-rich deposits (mines)



Challenges in radioecology: Linking radionuclide species in releases from sources to impact and risk assessment





RADIOACTIVE PARTICLES \rightarrow point sources of potential short- and <u>long</u>-term radioecological and analytical impact



- Transformation prosesses f(t)
 - Weathering rates and remobilisation
 - Underestimation of transfer factors for ecosystems and environmental effects in particle contaminated areas (change in speciation, Kd and CF)

DESTRUCTIVE TECHNIQUES



LEACHING EXPERIMENTS to estimate potential mobility and bioavailability

- Solubility in biologically relevant fluids (e.g. stomach juice)
- Solubility in rain water, sea water etc
- Sequential extractions

SOURCE IDENTIFICATION by radiometric methods and mass spectrometry

- Full dissolution/acid leaching
- Radiochemical separations
- Determination of activity concentrations and isotopic ratios in individual particles
- Information on isotope ratios which can be used as fingerprints for different sources
- MS techniques: ICP-MS, SIMS, AMS



Linking particle characteristics to potential bioavailability



U

- Low and moderate temperature events \rightarrow relatively high solubility
- Low solubility for particles from:
 - High temperature events under reducing conditions (non-oxidised or reduced U)
 - Oxidation of the Pu matrix to refractory oxides during fire
 - Vitrified soil or crystalline conglomerates of soil produced at extremely high temperatures (nuclear detonations)

Source: nm - µm sized particles Impact: Retention of µm-nm particles **Underestimated transport – retention pathways**



Radioactive particles retained in grazing goats, Incorporated in GI tissues



Severe skin dose about 30 mGy/hr, unevenly distributed **Microdosimetry** challenge



Dounrey particles given to Blue

Mussels as food, retained in the

gut and deposited in tissues

⁶⁰Co NPs uptake in reproductive organs of eathwurm Coutris et al, 2012







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