

The migration of radionuclides in the unsaturated zone and with groundwater

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Fieldcourse on Chornobyl fallout



CONTENT

Soil as the main source of radioactivity Factors that affect the mobility of radionuclides in soil Vertical distribution in soil profiles – case studies Migration of plutonium isotopes in ground water at the subsurface waste dump site Factors Controlling the Behaviour of Radionuclides in Soil



Soil properties

Mineralogical composition, pH, organic matter, humidity

Radionuclide elemental properties

(3H, 36Cl, 99Tc, 90Sr,137Cs, U, Pu Climate condition

Soil humidity, vegetation period

Different soil types and environmental conditions suggest different radionuclide behaviour between soils



Main processes affected the mobility of RN





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Modeling

MATPASS. Model of convective diffusion in an absorbing media

$$\begin{cases} \frac{\partial C_2(x,t)}{\partial t} = \frac{\partial}{\partial x} \left(D_2(x) \frac{\partial C_2(x,t)}{\partial x} - v(x) \cdot C_2(x,t) \right) + \alpha \cdot C_1(x,t) - K_s(x) \cdot C_2(x,t) - w(x) \cdot C_2(x,t) + \delta(x,0) \cdot C_p(t) \cdot R(t) \cdot P + K_{DS}(x) \cdot C_3(x,t) \\ -w(x) \cdot C_2(x,t) + \delta(x,0) \cdot C_p(t) \cdot R(t) \cdot P + K_{DS}(x) \cdot C_3(x,t) \\ \frac{\partial C_3(x,t)}{\partial t} = K_s(x) \cdot C_2(x,t) - K_{DS}(x) \cdot C_3(x,t) \\ \frac{\partial C_1(x,t)}{\partial t} = \frac{\partial}{\partial x} \left(D_1(x) \frac{\partial C_1(x,t)}{\partial x} \right) - \alpha C_1(x,t) \\ \frac{\partial C_p(t)}{\partial t} = \frac{1}{P \cdot H} \cdot \int_H w(x) \cdot C_2(x,t) \, dx - C_p(t) \cdot R(t) \end{cases}$$

x – depth, cm

t – time after deposition, year

H – depth of root inhabited layer, cm

v(x) – convective flux velocity, cm year⁻¹

 α - fuel particles destruction rate, year⁻¹

P – coefficient, kg cm⁻³

 $C_P(t)$ – concentration of radionuclide in plants, Bq kg⁻¹

 $C_1(x,t)$ – concentration of radionuclide in soil (FP component), Bq cm⁻³

 $C_2(x,t)$ and $C_3(x,t)$ – concentration and equilibrium concentration of radionuclide in soil (mobile form), Bq cm⁻³

 $C_3(x,t)$ – concentration of radionuclide in soil (absorbed form), Bq cm⁻³

 $C_1(0,0)=C_{10}; C_2(0,0)=C_{20}; C_3(0,0)=C_{30}; C_P(0)=0; C_1(x,0)=0; C_2(x,0)=0; C_3(x,0)=0$

 $C_1(\infty,t) = C_2(\infty,t) = C_3(\infty,t)=0;$ No flux of radionuclide through the ground surface

w(x) – radionuclide uptake by root system, year⁻¹ $K_S(x)$ – sorption coefficient, year⁻¹ $K_{DS}(x)$ – desorption coefficient, year⁻¹ $K_D(x)$ – distribution coefficient $D_1(x)$ – FP migration coefficient, cm² year⁻¹ $D_2(x)$ – diffusion coefficient of mobile form, cm² year⁻¹ R(t) – dynamics of radionuclides plant-to-soil transfer, year⁻¹.

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Determination of Kd of Sr: column experiments



Vertical migration of RNs(Chornobyl fallout)







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Tracer experiment in unsaturated zone (Beach site, 1999) COMET





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Vertical migration of RNs(Chornobyl fallout):



different elemental properties







Sep 2000, Rivne region,

Pasture;

Soil - Meadow light loamy

25-30

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Vertical migration of RNs(Chornobyl fallout):



different elemental properties



Vertical migration of RNs(Chornobyl fallout):



The experimental distribution of radionuclides in soil profile (sampling point Nº13, 1997

Vertical ¹³⁷Cs distribution in soil







15-20

Vertical ¹³⁷Cs distribution in soil at the site No 1COMET



Source characterization: case study

Trench #22:

Volume \approx 1300 m³ Radionuclides contents (2000): ¹³⁷Cs \approx 600 GBq ²³⁸Pu \approx 2.4 GBq ^{239,240}Pu \approx 4.3 GBq

General information:

•Distance from ChNPP: ~1.5 km W

•Contamination type: mainly fuel component

•History of the site: trees cut in 1987. Soil and trees buried in 1988 in the trenches

Activity buried: unknown

•GWL: ~ 3 m beneath the ground surface

•CPS: rectangular area 100x80 m containing the trench #22

Source term characterization (trench #22) COMET

⁹⁰Sr distribution under the trench

Source term characterization (trench #22) COMET

⁹⁰Sr distribution under the trench

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⁹⁰Sr distribution under the trench

Assessment of biogenic migration of radionuclides from the COMET trench

¹³⁷Cs and ⁹⁰Sr specific activities vertical distributions in upper 20 cm soil layer: a) under the birch (50 cm from trunk); b) under the birch (1.5 m from trunk); c) outside the area of leaves deposition (autumn 1998, at the trench surface). Activity, kBq/kg Activity, kBq/kg Activity, kBq/kg 100 1 10 10 100 1000 1 10 100 1000 0--2 0--2 0--2 Sr fixed 2--4 2--4 2--4 Sr fixed Sr exch. Sr exch. 4--6 4--6 4--6 🛛 Cs-137 🛛 Cs-137 6--8 6--8 6--8 8--10 8--10 8--10 Jepth, cm Depth, cm 10--12 10--12 ម្<u>ត</u> 10--12 Debt⁴ Debt⁴ 12--14 12--14 14-16 14-16 14-16 Sr fixed 16-18 16-18 16-18 Sr exch. 18-20 🛛 Cs-137 18-20 18-20 C) a) b)

Assessment of biogenic migration of radionuclides from the trench

¹³⁷Cs and ⁹⁰Sr specific activities vertical distributions in upper 40 cm soil layer: a) under the birch (1.2 m from trunk); b) under the pine (60 cm from trunk); c) outside the area of leaves and needles deposition (summer 1999, <u>at the trench surface</u>).

Assessment of biogenic migration of radionuclides from the COMET trench

Radionuclides transportation during one year (1999) from the trench to the ground surface

Plant	Number of	¹³⁷ Cs,	⁹⁰ Sr, MBq
	plants	MBq	
Pines	47	9.2	52
Birches	14	1.6	11
Bushes (I)	42	0.5	15
Bushes (II)	7	0.1	2
Grass	30 m ²	6.6	13
Total		18	183

Conclusions

♦ ⁹⁰Sr exchangeable forms can be concentrated in upper soil layer under the crowns of trees growing at the trench.

♦ approximately 45 kBq of ¹³⁷Cs and 300 kBq of ⁹⁰Sr are deposited at 1 m² of the trench surface each year.

◆ ¹³⁷Cs content in timber at the trench in 1999 was about 20 MBq, ⁹⁰Sr – 180 MBq.

Monitoring well layout at Chernobyl Pilot Site

COMET

Plutonium behavior in the Environment

Complicate chemistry (several oxidation states)
High tendency for sorption and complexation
In general, low mobility in the Environment
Usually, migration of Pu from waste storages or from contaminated soil had been thought unimportant because of its high absorption by soil

In the present time there are evidences of plutonium high mobility in aquifer
 Mechanism of Pu migration is not known enough for long-time prediction
 In hundreds years radiological situation will be determined by THE

Main questions

- Spatial plutonium distribution in aquifer at present ?
- Mechanism of plutonium migration from the trench into aquifer?
 - Distribution of size fractions of Pu in aquifer?
 - fast migration of plutonium predecessors and their decay downstream the source (mainly ²⁴⁴Cm to ²⁴⁰Pu)?
 - Chemical compound with weights less than 5kDa??

Main experimental activity:

Plutonium spatial distribution in aquifer

□ Size distribution of plutonium

Estimation of in-situ Pu distribution coefficients

for eolian sand (saturated zone)

for waste material (trench body, unsuturated zone)

Plutonium spatial distribution in aquifer along WS profile (2005, 2006)

The trench is main source of Pu in aquifer

Plutonium spatial distribution in aquifer along lab profile (2005,2006)

Maximum of Pu activity is located at the distance of 8-10 from the center of the trench in upper part of aquifer

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Ratio ^{239,240}Pu/²³⁸Pu in ground water samples COMET

Fast migration of plutonium predecessors is not the main cause of such plutonium spatial distribution

The size fractionation data on plutonium activities in collected colloids suggest that significant part of Pu (50-80%) passes through the 5 kDa filter (several nm).

The distribution of plutonium activity between fractions remained stable through time when analyzed in different samples from the same piezometer.

Ultrafiltration

Colloidal plutonium fractions distribution

Plutonium species in the aquifer were two orders of magnitude more mobile than in the trench

Conclusions

All plutonium in the aquifer originated from the trench

The obtained spatial distribution of plutonium in the aquifer appears to be highly directional, reflecting the orientation of the hydraulic gradient. Lateral migration of the radionuclides occurs mainly in the eolian sediments layer, which is characterized by high permeability and low CEC.

The obtained Pu K_d value in the aquifer ($n \cdot 10 \text{ L kg}^{-1}$) is two orders of magnitude less than those in the source material ($n \cdot 10^3 \text{ L kg}^{-1}$)

□Obtained size fractionation data suggest that a significant part of plutonium (50-98 %) in the groundwater sampled close to the source is associated with the very low molecular weight fraction (< 1 kDa).

